Progress of the Suzuki-Miyaura Reaction 叶青杨 3210100360

(Department of Chemistry, Zhejiang University, Hangzhou, Zhejiang, 310058)

Abstract Due to cheapness and earth abundance, the Suzuki-Miyaura cross-coupling reaction has gradually shifted from mainly using the traditional Pd metal to using the more attractive first-row transition metal Ni in recent years. In this review, we briefly summarize some progress in Pd and Ni catalyzed Suzuki-Miyaura reactions.

Keywords Suzuki-Miyaura Reaction; C-C Coupling; Abundant Transition Metals; Pd Catalyst; Ni Catalyst

1 Introduction

In organometallic chemistry, coupling reactions related to the formation of carbon-carbon single bonds are particularly noteworthy. The Suzuki-Miyaura reaction, as an important one among them, has developed very rapidly in recent years. In this review, we will provide a systematic introduction to the Suzuki-Miyaura reaction catalyzed by Pd and Ni.

> R^{1} -B(R)₂ + R²-X R = 烷基、羟基、烷氧基 R¹ = 烷基、烯基、烷氧基 R¹ = 烷基、烯基、烷基、芳基等 R² = 烷基、烯基、芳基等 X = F, Cl, Br, I, OMs, OTs, OP(O)(OR)₂,OC(O)(OR)₂,OC(O)(NR₂)₂等

Figure 1: Typical Conditions for the Suzuki-Miyaura Cross-Coupling Reaction

The Suzuki-Miyaura reaction is commonly defined as a reaction forming carbon-carbon bonds between organoboron compounds and organic halides or pseudo-halides, such as sulfonates, phosphates, carbonates, etc., under the action of transition metals like palladium or nickel, as shown in Figure 1. This reaction was first reported in 1979 by the Japanese organic chemists A. Suzuki and N. Miyaura. The appeal of this reaction lies in its mild reaction conditions, broad functional group tolerance, and good stereo- and regioselectivity. Organoboron reagents are readily available and have considerable stability against air and moisture. Moreover, the boronic acid by-products of the reaction are usually less toxic or even non-toxic and easily separable due to their water solubility. Thus, upon its discovery, it quickly became an attractive solution for many organic syntheses.^{1,2}

As shown in Figure 2, the mechanism of the Suzuki-Miyaura reaction usually follows anM⁽⁰⁾/M^(II) mechanism $_{\circ}$ T he active catalytic intermediate, a zero-valent transition metal complex M₁⁽⁰⁾ compound A, is obtained by reducing the initial divalent transition metal M₁^(II) complex. Compound A undergoes oxidative addition with organic halides or pseudo-halides to form an M₁^(II) intermediate B.X⁻ in B can exchange with Y⁻ to form C, and B or C undergoes transmetalation to form D. Finally, a reductive elimination reaction occurs, producing the product and regenerating the zero-valent M₁⁽⁰⁾ complex A, completing the catalytic cycle. The oxidative addition of the halide is the rate-determining step, and the reaction rate typically follows I≫Br>OTf≫Cl.³

Many studies have found that under certain conditions, by adjusting the structural properties of ligands, Ni catalysts can catalyze the Suzuki-Miyaura reaction via a $Ni^{(I)}/Ni^{(III)}$ process, but its mechanism differs from that shown in Figure 2, as illustrated in Figure 3. The zero-valent nickel complex undergoes a single electron transfer reaction to form the $Ni^{(I)}$ intermediate E. E undergoes transmetalation to form the $Ni^{(I)}$ intermediate F. F continues to undergo oxidative addition with the halide to yield the $Ni^{(III)}$ intermediate G. Finally, after a reductive elimination, the product is formed and the $Ni^{(I)}$ complex E is regenerated, completing the catalytic cycle of the reaction.



Figure 2: Common Mechanism of the Suzuki- Figure 3: The Ni^(I)/Ni^(II) Mechanism of theMiyaura Cross-Coupling ReactionSuzuki-Miyaura Cross-Coupling Reaction

2 Pd-Catalyzed Suzuki-Miyaura Reaction

2.1 Reaction Examples Using 9-BBN⁴



Figure 4: Suzuki-Miyaura Coupling Reaction Figure 5: Suzuki-Miyaura Coupling Reaction with Steroidal Moiety⁵ with Electron-Rich, Hindered Aryl Bromides⁶



Figure 6: Application of the Suzuki-Miyaura Figure 7: Application of the Suzuki-Miyaura Coupling Reaction in Amino Acid Synthesis⁷ Coupling Reaction in Amino Acid Synthesis⁸



Figure 8: Chiral Chemistry of the Suzuki-Miyaura Coupling Reaction⁹

2.2 Reaction Examples with Phosphate Esters



Figure 9: Suzuki-Miyaura Coupling Reaction Figure 10: Suzuki-Miyaura Coupling Reaction with Vinyl Phosphate Esters¹⁰ with Unactivated Vinyl Phosphate Esters¹¹

2.3 Reaction Examples with p-Toluenesulfonate Esters



Figure 11: Suzuki-Miyaura Coupling Reaction Figure 12: Suzuki-Miyaura Coupling Reaction with p-Toluenesulfonate Esters in the Presence with p-Toluenesulfonate Esters in the Presence of X-Phos¹² of CM-Phos¹³

2.4 Reaction Examples with Methylsulfonate Esters



Figure 13: Suzuki-Miyaura Coupling Reaction Figure 14: Suzuki-Miyaura Coupling Reaction with Methylsulfonate Esters in the Presence of with Methylsulfonate Esters in the Presence of CM-Phos¹⁴ BrettPhos¹⁵

Examples of Pd-catalyzed Suzuki-Miyaura coupling reactions are fewer compared to those catalyzed by Ni, and in recent years, organic chemists have primarily focused their interest on the Suzuki-Miyaura coupling reactions involving Ni. Moreover, due to the limited studies on the mechanisms of Pd-catalyzed reactions and the narrow applicability for sp³-hybridized and bulky, nonactivated sp²-hybridized halides/pseudo-halides, as well as related asymmetric coupling reactions, the scope remains somewhat limited. Consequently, in the research on Ni-catalyzed Suzuki-Miyaura coupling reactions, there are more examples.

3 Ni-Catalyzed Suzuki-Miyaura Reaction^{16,17}

3.1 Reaction Examples with Aryl Halides



Figure 15: Suzuki-Miyaura Coupling Reaction Figure 16: Suzuki-Miyaura Coupling Reaction with Aryl Bromides¹⁸ with Aryl Chlorides¹⁸

3.2 **Reaction Examples with Alkyl Halides**



Figure 17: Suzuki-Miyaura Coupling Reaction Figure 18: Suzuki-Miyaura Coupling Reaction with Alkyl Chlorides²⁰ with Alkyl Bromides and Iodides¹⁹

3.3 **Reaction Examples with Arylsulfonamides**



pling Reactions with Arylsulfonamides²¹

Figure 19: High-Yield Suzuki-Miyaura Cou-Figure 20: Examples of Suzuki-Miyaura Coupling Reactions with Arylsulfonamides In Synthesis (Flurbiprofen)²¹

3.4 Reaction Examples with Aryl Methyl Ethers



Figure 21: Suzuki-Miyaura Coupling Reaction with Aryl Methyl Ethers²²

3.5 Reaction Examples with Phosphate Esters





3.6 Reaction Examples with Aryl Carbonates



Figure 23: Suzuki-Miyaura Coupling Reaction with Aryl Carbonates^{26–28}



Figure 24: Examples of Suzuki-Miyaura Coupling Reactions with Aryl Carbonates in Synthesis^{26,29}

3.7 Reaction Examples with Aminosulfonate Esters



Figure 25: Examples of Suzuki-Miyaura Coupling Reactions with Carbamate Esters in Synthesis^{21,30}

3.8 Reaction Examples with Aminosulfonate Esters



Figure 26: Suzuki-Miyaura Coupling Reactions of Aminosulfonate Esters under Various Ni Catalysts^{31–33}

3.9 Reaction Examples with Aryl Fluorides



Figure 27: Suzuki-Miyaura Coupling Reaction with Aryl Fluorides^{33,34}

4 Application of Suzuki-Miyaura Coupling Reaction in the Total Synthesis of Complex Natural Products

In recent years, with the advancement of organic synthetic chemistry, more cutting-edge content in organometallic chemistry has been applied in practical synthesis. The Suzuki-Miyaura coupling reaction is playing an increasingly important role in the efficient total synthesis of many complex natural products.

In 2014, the research group led by Tang Wenjun³⁵ developed a mild and efficient method for asymmetric Suzuki-Miyaura cross-coupling reactions (Figure 28). Using a chiral catalyst system composed of $Pd(OAc)_2$ and the chiral phosphine ligand L2, they achieved asymmetric cross-coupling of bulky aryl bromides with boronic acids. The reaction yielded P-configuration products with up to 96% yield and 93% ee. Through subsequent transformations using this intermediate, the first to-tal synthesis of korupensamine A was realized. Similarly, using ent – L2 as the chiral ligand, the corresponding M-configuration products were obtained with high enantioselectivity and yield, leading to the first total synthesis of korupensamine B. Later, the authors employed the Suzuki-Miyaura cross-coupling reaction to couple reaction intermediates, completing the efficient total synthesis of the polyaromatic natural product michellamine B.

4.1 Total Synthesis of korupensamine A, korupensamine B and michellamine B



Conditions: (a) TBSCl, TEA, DCM, -78 °C to rt, 59%; (b) NBS, DCM, -15 °C, 86%; (c) BOPCl, TEA, DCM, 98%; (d) 14, Pd(OAc)₂/L2, K₃PO₄, toluene/H₂O = 5/1, 96%, 93% ee; (e) NaOH, MeOH, 97%; (f) BnBr, K₂CO₃, DMF, 95%; (g) EtNO₂, HO(CH₂)-NH₃⁺HCO₂⁻, 60 °C, 92%; (h) Fe, AcOH, Ac₂O, DMF, 70 °C, 60%; (i) 1 mol % Rh(nbd)₂BF₄/L4, H₂ (20 atm), DCM, 0 °C, 12 h, 99%, d.r. 92/8; (j) BnBr, NaH, DMF, 96%; (k) POCl₃, 2,4,6-collidine, toluene, 100 °C, 4 h; (l) NaBH₄, EtOH, -78 °C to rt, 85% (over 2 steps, *trans:cis* = 6/1); (m) Pd/C (5%), H₂ (1 atm), MeOH/DCM = 2/1, 3 h, 95%.



Figure 28: Total Synthesis of korupensamine A, korupensamine B, and michellamine B

4.2 Diverse Total Synthesis of Bird's Nest Alkane and Hamigeran Series of Natural Products

In 2018, the research group led by Han Fusha^{36,37} developed a new class of cyclic palladium catalysts based on phosphoramidite ligands. These catalysts exhibited high catalytic activity for Suzuki-Miyaura cross-coupling reactions between bulky, non-activated cyclic trifluoromethanesul-fonate enol ester derivatives and arylboronic acids (esters). The reaction could proceed smoothly at room temperature. Additionally, this reaction demonstrated broad substrate adaptability, yielding high yields of coupled products for variously substituted five-membered, six-membered, and bridged ring enol derivatives. Further studies showed that this method played a decisive role in the straightforward, efficient, and diverse total synthesis of diterpene natural products, including the bird's nest alkane type and hamigeran.



^cThe reaction was run at rt for 5 h ^dThe reaction was performed with 6.6 g of **15a** and was quenched until **15a** had disappeared as monitored by TLC

Figure 29: Condition Screening for the Synthesis of Key Intermediates

5 Conclusion and Outlook

In summary, this review has provided a comprehensive overview of the Suzuki-Miyaura cross-coupling reaction, focusing on the advancements and applications of both palladium- and nickel-catalyzed methods. The evolution of these reactions over recent years demonstrates their versatility and growing importance in the field of synthetic chemistry. Particular attention has been given to the nickel-catalyzed Suzuki-Miyaura reaction due to its increasing prominence and potential. The future of this methodology appears promising, with expectations of continued innovation and broader application in complex molecule synthesis. We anticipate further developments in this area, potentially leading to more efficient, sustainable, and cost-effective synthetic strategies.

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